

Dihydridotetrakis(4-picoline-*N*)silicon dibromide chloroform hexasolvate

Michael Bolte^{a*} and Alexander Faber^b

^aInstitut für Organische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Straße 11, 60439 Frankfurt/Main, Germany, and ^bInstitut für Physikalische und Theoretische Chemie, J. W. Goethe-Universität Frankfurt, Marie-Curie-Straße 11, 60439 Frankfurt/Main, Germany

Correspondence e-mail: bolte@chemie.uni-frankfurt.de

Key indicators

Single-crystal X-ray study
 $T = 173$ K
 Mean $\sigma(\text{C}-\text{C}) = 0.008$ Å
 R factor = 0.054
 wR factor = 0.148
 Data-to-parameter ratio = 18.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $[\text{SiH}_2(\text{C}_6\text{H}_7\text{N})_4]\text{Br}_2 \cdot 6\text{CHCl}_3$ or $\text{C}_{24}\text{H}_{30}\text{N}_4\text{Si}^{2+} \cdot 2\text{Br}^- \cdot 6\text{CHCl}_3$, contains a hexacoordinated Si atom located on a crystallographic centre of inversion. The coordination of the Si atom can be described as a slightly distorted octahedron, with the 4-picoline ligands in the equatorial plane and the two H atoms occupying axial positions. The title compound is isomorphous with its analogue where the Br^- ions are substituted by Cl^- ions.

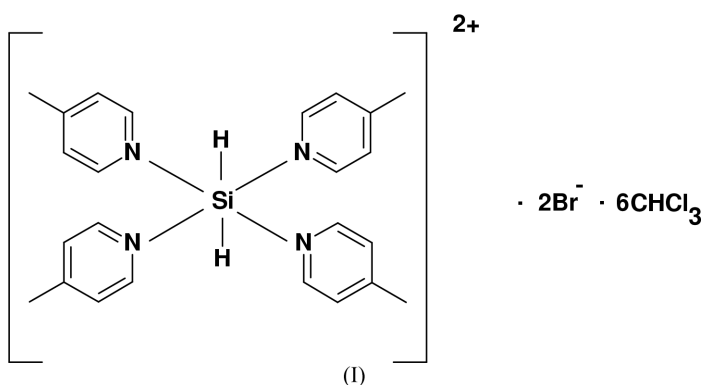
Received 23 January 2001

Accepted 31 January 2001

Online 13 February 2001

Comment

The title compound, (I), is isomorphous with its analogue where the Br^- ions are substituted by Cl^- ions (Stumpf & Bolte, 2001).



Experimental

The title compound was prepared as described by Faber (2000).

Crystal data

$\text{C}_{24}\text{H}_{30}\text{N}_4\text{Si}^{2+} \cdot 2\text{Br}^- \cdot 6\text{CHCl}_3$
 $M_r = 1278.64$
 Triclinic, $P\bar{1}$
 $a = 9.506$ (1) Å
 $b = 11.337$ (2) Å
 $c = 12.960$ (2) Å
 $\alpha = 92.28$ (1)°
 $\beta = 104.06$ (1)°
 $\gamma = 98.20$ (1)°
 $V = 1337.0$ (3) Å³

$Z = 1$
 $D_x = 1.588$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 8192 reflections
 $\theta = 1-25^\circ$
 $\mu = 2.47$ mm⁻¹
 $T = 173$ (2) K
 Needle, colourless
 $0.6 \times 0.4 \times 0.3$ mm

Data collection

Siemens CCD three-circle diffractometer
 ω scans
 Absorption correction: empirical (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.290$, $T_{\max} = 0.477$
 15 431 measured reflections
 4721 independent reflections
 3954 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.027$
 $\theta_{\text{max}} = 25.0^\circ$
 $h = -11 \rightarrow 8$
 $k = -13 \rightarrow 13$
 $l = -15 \rightarrow 15$
 98 standard reflections
 frequency: 1440 min
 intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.148$
 $S = 1.03$
 4721 reflections
 256 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0746P)^2 + 3.3546P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.020$
 $\Delta\rho_{\max} = 1.98 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.75 \text{ e } \text{\AA}^{-3}$

Table 1

 Selected geometric parameters (\AA , $^\circ$).

Si1—N1	1.958 (4)	Si1—N2	1.959 (5)
H1Si—Si1—N2	91 (2)	N1—Si1—N2	89.79 (17)

All H atoms bonded to C atoms were located by difference Fourier synthesis and refined with fixed individual displacement parameters [$U(\text{H}) = 1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$ or $U(\text{H}) = 1.2U_{\text{eq}}(\text{C})$] using a riding model with $\text{C—H}(\text{aromatic}) = 0.95$, $\text{C—H}(\text{methyl}) = 0.98$ or $\text{C—H}(\text{tertiary}) = 1.00 \text{ \AA}$. The methyl groups attached to the aromatic rings were allowed to rotate about their local threefold axis. The H atom bonded to Si was refined isotropically applying a restraint of $1.40 (1) \text{ \AA}$ to the Si—H distance

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97*.

References

- Faber, A. (2000). PhD thesis, University of Frankfurt, Germany.
 Sheldrick, G. M. (1990). *Acta Cryst. A* **46**, 467–473.
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
 Siemens (1995). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Stumpf, T. & Bolte, M. (2001). *Acta Cryst. E* **57**, o209–211.

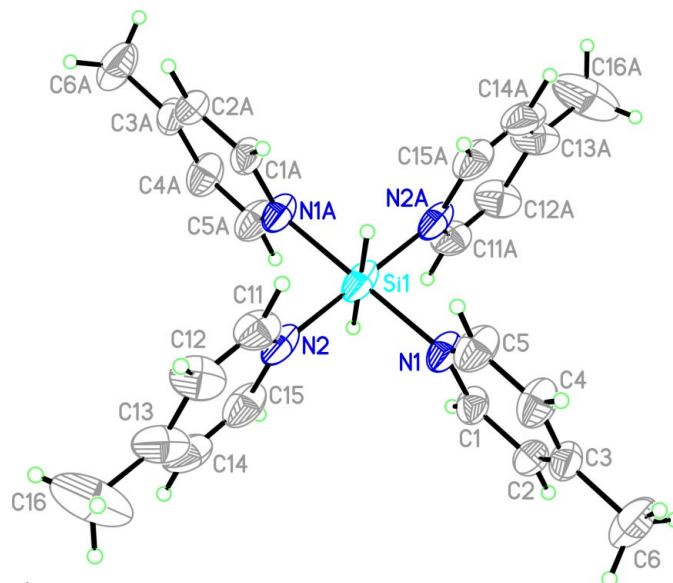


Figure 1
 A perspective view of the cation of (I) with the atom-numbering scheme. Displacement ellipsoids are at the 50% probability level and H atoms are drawn as small spheres of arbitrary radii.

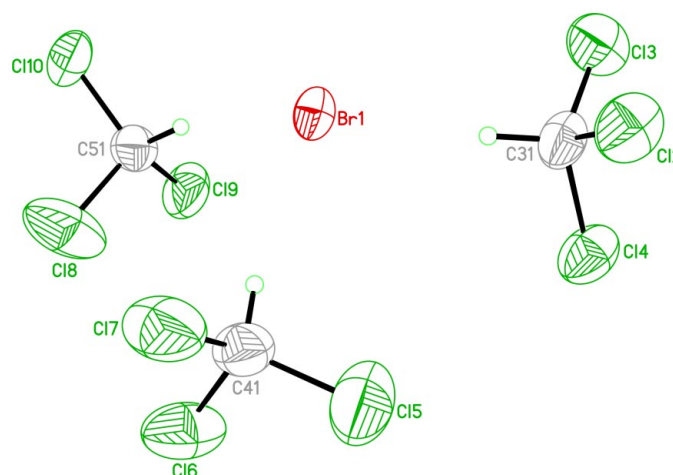


Figure 2
 A perspective view of the anion and the solvent molecules of (I) with the atom-numbering scheme. Displacement ellipsoids are at the 50% probability level and H atoms are drawn as small spheres of arbitrary radii.